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Enantioselective Total Synthesis of (+)-Jasplakinolide

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ABSTRACT

An enantioselective total synthesis of (+)-jasplakinolide is described. The synthesis of the polyketide template utilized a diastereoselective syn-aldol, ortho-ester Claisen rearrangement followed by efficient conversion to a cyanide. The β -amino acid unit was constructed in a highly diastereoselective manner utilizing nucleophilic addition to a chiral sulfinimine. Yamaguchi macrocyclization and removal of the protecting group provided a convenient access to (+)-jasplakinolide.

Jasplakinolide (1), a 19-membered cyclic depsipeptide, initially was isolated from the marine sponge *Jaspis splendens* in 1986.¹ It was later found in other marine sponges, including *Auletta* sp., *H. minor*, and *Cymbastela* sp.² Jasplakinolide exhibited a number of very interesting biological properties. It is active against 36 human solid tumor types in cell culture assays.³ It has also exhibited other important biological properties, including insecticidal, antifungal, and antihelminthic activities.⁴ The mechanism of action is known

to involve stabilization of actin filaments by binding to F-actin similar to phalloidin.⁵ Preclinical trials of jasplakinolide were carried out by the National Cancer Institute as an anti-actin agent. However, the study was terminated as it showed significant toxicity.⁶ Jasplakinolide is often used as a molecular probe for actin polymerization studies. Its biological properties and structural features attracted attention for total synthesis⁷ and structural modification.⁸ We recently reported an enantioselective total synthesis of (—)-doliculide,

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another cyclic depsipeptide isolated from *Dolabella auricularia* by Ishiwata and co-workers. Using synthetic doliculide, we established that it too arrests cell growth at the G_2/M phase of the cell cycle by interfering with normal actin assembly similar to jasplakinolide. Doliculide exhibited no toxicity. Structural similarities of doliculide and jasplakinolide suggest that both compounds bind to the same site on F-actin. To investigate structure—activity studies of jasplakinolide, we sought an efficient synthesis of jasplakinolide for further structural modification. Herein we report an enantioselective total synthesis of (+)-jasplakinolide. The synthesis features asymmetric synthesis of 8-hydroxynonenoic acid 3 and (R)- β -tyrosine unit 4.

Our convergent approach to jasplakinolide synthesis is shown in Figure 1. As shown, 1 is derived from hydroxyl

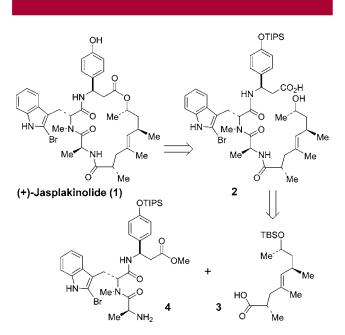


Figure 1. Retrosynthetic analysis of (+)-jasplakinolide.

acid **2** by a macrocyclization reaction. Macrocyclic precursor **2** would be obtained by coupling of the polypropionic acid segment **3** and tripeptide **4**. The tripeptide fragment is composed of amino acids, (R)-2-bromoabrine, (R)- β -tyrosine, and (S)-alanine.

The synthesis of protected 8-hydroxynonenoic acid **3** is shown in Scheme 1. Asymmetric aldol addition of the boron enolate derived from oxazolidinone **5** with methacrolein gave the aldol adduct **6** in 94% yield. Allylic alcohol **6** was subjected to acid-catalyzed ortho ester Claisen rearrangement with triethyl orthopropionate at 140 °C for 1 h to afford γ , unsaturated ester **8** in 93% yield as a mixture (7:1 by ¹H NMR) of diastereomers. This Claisen rearrangement protocol di-

Synthesis of Carboxylic Acid 3 Scheme 1. nBu₂BOTf, iPr₂NEt CH₂Cl₂, 0 °C Methacrolein -78 °C to 0 °C 94% 6 EtC(OEt)3, EtCO₂H (cat.) 93% 140 °C **OEt** OEt Мe Йe (7:1)7 1. LiBH₄, EtOH 2. Ph₃P, DIAD Me₂C(OH)CN 1. Ra-Ni, NaH₂PO₃, Pyr, aq AcOH Йe Йe Мe 10 2. MeMgBr, THF Me 73% 1. Ph₃P, DIAD 4-nitrobenzoic acid, Et₂O 1. TBSOTf, Et₃N, 2. K2CO3, EtOH CH₂Cl₂ 78% 2. aq LiOH, Мe Йe THF, MeOH OH 11 93%

astereoselectively introduced the α-methyl carbonyl functionality through 1,4-chirality transfer. 12 The Claisen rearrangement presumably proceeded through a chair-like transition state as shown in 7 that accounts for the observed diastereoselectivity as well as the E-olefin geometry in 8. Ethyl ester 8 was converted to cyanide 9 by reduction with LiBH4 and EtOH¹³ followed by Mitsunobu reaction of the resulting alcohol using acetone cyanohydrin in the presence of diisopropyl azodicarboxylate and triphenylphosphine.¹⁴ Nitrile **9** was obtained in 62% yield in a two-step sequence. Reduction of nitrile with Raney nickel in the presence of NaH₂PO₃, pyridine, and aqueous acetic acid provided the corresponding aldehyde. 15 Reaction of aldehyde with methylmagnesium bromide afforded diastereomeric alcohols 10 and 11 as a 1:1 mixture in 73% yield in two steps. The alcohols were separated by flash column chromatography over silica gel. Mitsunobu inversion 16 of (R)-alcohol 10 with Ph₃P and p-NO₂-benzoic acid in the presence of diisopropyl azodicarboxylate followed by ester hydrolysis of the resulting benzoate derivative with potassium carbonate in ethanol furnished the desired (S)alcohol 11. Protection of alcohol 11 with TBSOTf, Et₃N at 23 °C, and saponification with aqueous lithium hydroxide furnished 8-hydroxynonenoic acid 3 in 93% yield.

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Diastereoselective synthesis of desired (R)- β -tyrosine derivative was achieved using an asymmetric protocol developed by Davis and co-workers, ¹⁷ as shown in Scheme 2. Aldehyde **12** was readily prepared by protection of

Synthesis of β -Amino Acid Units **14** and **4** Scheme 2. Ś. Ti(OEt)₄, CH₂Cl₂ (S)-p-toluene-13 TIPSO sulfinamide, 40 °C TIPSO 87% 1. NaHMDS, Et₂O 12 65% MeCO₂Me, -78 °C 2. TFA, MeOH, 0 °C NH₂ O OH OMe 15 14 **TIPSO** Me·N 1. 15, DCC, Me NΗ THF, 0 °C BOC TFA, CH₂Cl₂

4-hydroxybenzaldehyde with TIPSOTf and triethylamine in CH_2Cl_2 at 0 °C for 3 h. Reaction of **12** with enantiopure p-toluenesulfinamide in CH_2Cl_2 in the presence of $Ti(OEt)_4$ at 40 °C afforded sulfinimine **13** in 87% yield. Reaction of **13** with enolate derived from methyl acetate in diethyl ether at -78 °C afforded the corresponding addition product as a single diastereomer (by ¹H NMR). Treatment of this resulting sulfinamide with trifluoroacetic acid in CH_2Cl_2 removed the N-sulfinyl auxiliary and afforded (R)- β -tyrosine derivative **14** in 65% yield in two steps. Construction of the tripeptide fragment **4** was then achieved by coupling reaction of known dipeptide **15**^{7b} with **14** in the presence of DCC in THF at 0 °C for 12 h. Removal of the BOC group by exposure to trifluoroacetic acid in CH_2Cl_2 provided amino ester **4** in 81% yield.

The final assembly of jasplakinolide fragments is illustrated in Scheme 3. The coupling of the above amino ester 4 with 8-hydroxynonenoic acid 3 in the presence of DCC and HOBT in THF at -20 to 0 °C for 24 h furnished coupling product 16 in 80% yield. Saponification of 16 with aqueous LiOH at 23 °C affected removal of the TIPS group and afforded the corresponding phenolic acid. Treatment of the resulting phenolic acid with TIPSOTf in the presence of 2,6-lutidine at 23 °C for 2 h followed by exposure of the resulting TIPS derivative with aqueous potassium carbonate afforded seco acid 2 in 77% yield in three steps. Acid 2 was subjected to Yamaguchi macrolactonization protocol 18 with 2,4,6-trichlorobenzoyl chloride in the presence of DMAP to

Scheme 3. Synthesis of (+)-Jasplakinolide

provide the corresponding macrolactone in 82% yield. Removal of the TIPS group by treatment of the macrolactone with TBAF in THF at 0 °C for 10 min furnished synthetic (+)-jasplakinolide (1, $[\alpha]^{23}_D$ +67.7, c 0.2, CH₂Cl₂). The spectral data (1 H and 13 C NMR) of synthetic (+)-jasplakinolide are identical with those reported for the natural (+)-jasplakinolide.

In summary, we have achieved an enantioselective synthesis of (+)-jasplakinolide (1). The convergent synthesis features diastereoselective *syn*-aldol, ortho-ester Claisen rearrangement and asymmetric synthesis of the (R)- β -tyrosine derivative. The synthesis will provide a convenient access to a variety of jasplakinolide derivatives. Structural modifications of jasplakinolide are currently in progress.

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Supporting Information Available: Experimental procedures, spectral data for compounds 1–4, 6, 8–11, 13, 14, and 16, and ¹H NMR and ¹³C NMR spectra for compounds 1, 2, 4, 8–11, 13, 14, and 16. This material is available free of charge via the Internet at http://pubs.acs.org.

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